

## Chapter 12 Assessment Chemistry Answers

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### Chapter 12 Assessment Chemistry Answers

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### Chapter 12 Assessment Chemistry Answers

Solutions Manual Chemistry: Matter and Change • Chapter 12 239 CHAPTER 12 SOLUTIONS MANUAL Section 12.3 Liquids and Solids pages 415–424 Section 12.3 Assessment page 424 18. Contrast the arrangement of particles in solids and liquids. The particles are closer together in solids than in liquids because of intermolecular attractions.

### States of Matter

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Chapter 12 - Stoichiometry - 12 Assessment - Page 413: 64. Answer.  $1.78 \times 10^{25}$  molecules H<sub>2</sub>. Work Step by Step. We have amount methane and need molecules of hydrogen, so we use molar ratio of moles hydrogen to moles methane. Get the molar mass of methane by adding  $1 \times$  molar mass carbon +  $4 \times$  molar mass of hydrogen.

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CHAPTER 3 SOLUTIONS MANUAL Matter—Properties and Changes Matter—Properties and Changes Solutions Manual Chemistry: Matter and Change • Chapter 3 35 Section 3.1 Properties of Matter pages 70–75 Problem-Solving Lab 1. Explain why the flow of a compressed gas must be controlled for

practical and safe use.

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